

**1) (10 points)** You are travelling to Alpha Centauri in a star ship that is moving at  $0.80 c$  relative to Earth. Your twin brother is also headed to Alpha Centauri, in a second star ship moving at  $0.90 c$  relative to Earth and directly following you. As it happens, both of your watches were exactly synchronized when your twin brother's star ship was exactly one light-hour distant (behind yours) from your point of view. When your brother's star ship pulls even with yours, by how much will the times on the two watches differ?

**Solution**

We first need to calculate how fast your brother is moving relative to you. Since your brother is following you, we must subtract the given velocities like so:  $v = (0.9 c - 0.8 c) / [1 - (0.8)(0.9)] = 0.3571 c$ .

Thus, from your point of view, your brother will cover the light-hour distance between you and him in  $t = d/v = 1 \text{ c-hr} / 0.3571 c = 2.8 \text{ hours}$ . However, the time that will run off his watch must be dilated from your point of view into 2.8 hours. In other words, we have  $2.8 \text{ hrs} = \Delta t / (1 - 0.3571^2)^{1/2}$ , where  $\Delta t$  is the "slow" time that you see on his watch. We have  $\Delta t = (2.8)(0.934) = 2.615 \text{ hrs}$ . The difference between the two watches is  $2.8 - 2.615 = 0.185 \text{ hr} = 11.1 \text{ min}$ .

**2) (10 points)** A soap film with an index of refraction  $n = 1.40$  is hanging suspended in air. White light is shining on the film, but only light that has  $\lambda = 602$  nm is being transmitted through the film with 100% efficiency. What is the minimum thickness of the film? **Note** – to obtain full credit on this problem, you must indicate your logic. A bare equation will not suffice.

**Solution**

For 100% of the light to shine through the film, there must be no light reflected at that wavelength. In other words, we want the reflected waves to be fully destructive. There will be a phase flip at the first interface ( $1.00 < 1.40$ ), but there will not be a phase flip at the second interface ( $1.40 > 1.00$ ). So, if the two reflected waves are  $2\pi$  out of phase, they will destructively interfere when one includes the phase flip.

We have  $\delta = 2\pi \Delta x/\lambda$ . In this case  $\Delta x = 2L$ , where  $L$  is the thickness of the soap film, and  $\lambda = \lambda_{\text{air}}/n$ . Inserting numbers,  $2\pi = 2\pi (2L)(1.4)/602$  nm, or  $L = (602 \text{ nm})/2.8 = 215$  nm.

**3) (10 points)** The muon is an unstable subatomic particle which has a mass of  $105.66 \text{ MeV}/c^2$  and a lifetime of  $2.197 \times 10^{-6} \text{ s}$ . Suppose you created a beam of muons here in Evanston and fired them through an underground tunnel to Los Angeles, 3200 km away. What is about the minimum kinetic energy the muons would need to have to reach Los Angeles?

**Solution**

This problem calls mostly for logic, not high mathematics. Even moving at the speed of light, a muon can only travel  $(3 \times 10^8)(2.197 \times 10^{-6}) = 659.1 \text{ m}$  within its lifetime, which is a bit short of 3200 km. So, we see that we must time-dilate its lifetime by at least  $3,200,000 / 659.1 = 4855$  times to reach LA. Since the relativistic time-dilation formula and  $E = mc^2$  use the same  $\gamma$ -factor, this means that we must raise its total energy to at least  $(4855)(105.66) = 513 \text{ GeV}$ . Now,  $E_K = E_{\text{total}} - m_0$ , but 513 GeV is  $\gg 106 \text{ MeV}$ , so we can neglect the rest mass, and thus  $E_K \approx 513 \text{ GeV}$ .

P.S. -- With a  $\gamma$ -factor of 4855, one does not need to worry about our approximation that the muon is moving at the speed of light. If you care to calculate it out,  $\gamma = 4855$  corresponds to  $v = 0.999999979 c$ .

4) The radius of a hydrogen atom, also known as the Bohr Radius, is about  $5.29 \times 10^{-11}$  m. The minimum wavelength of E&M radiation that a real hydrogen atom can emit is about 91 nm. The maximum is almost infinitely long. But, suppose you were to model this atom as an infinitely deep one-dimensional box.

4a) (4 points) What minimum wavelength of E&M radiation could the box emit? Briefly explain why.

**Solution**

Since the box is infinitely deep, it can have energy jumps that are infinitely high, that is, with infinite energy. But infinite energy corresponds to zero wavelength, so the box effectively has no minimum wavelength.

4b) (6 points) Calculate the maximum wavelength of E&M radiation the box could emit. (The mass of the electron is  $9.11 \times 10^{-31}$  kg.)

**Solution**

The maximum wavelength will be given by the smallest energy jump, which will be from  $n = 2$  to  $n = 1$ . We have  $E = n^2 h^2 / 8mL^2 = (1)^2 (6.626 \times 10^{-34})^2 / 8(9.11 \times 10^{-31})(4)(5.29 \times 10^{-11})^2 = 33.6$  eV for the first energy level in the box. The second energy level will be  $(2)^2(33.6) = 134.4$  eV. The difference in energy is thus  $134.4 - 33.6 = 100.8$  eV. The corresponding wavelength is  $\lambda = 1240 / 100.8 = 12.3$  nm.

**5) (10 points)** You have a solar cell which is 15 cm long and 5 cm wide. The cell is made of a material which has a work function of 2.1 eV. Bright noon-time sunlight is shining on the cell, and the cell is producing 1.3 W of electric power. What is the efficiency of the solar cell, in terms of electrons produced per input photon?

Intensity of bright sunlight at noon:  $1300 \text{ W/m}^2$

Wavelength of sunlight you can assume:  $501 \text{ nm}$

$1 \text{ amp} = 6.24 \times 10^{18} \text{ electrons / s}$

Electric Power,  $P = IV$

**Solution**

The power hitting the cell is  $(1300)(0.15)(0.05) = 9.75 \text{ W}$ . The energy of the incoming photons is  $E = 1240 / 501 = 2.475 \text{ eV} = 3.965 \times 10^{-19} \text{ J}$ , so that means we have  $9.75 / (3.965 \times 10^{-19}) = 2.46 \times 10^{19}$  photons/s striking the cell.

Now, the energy of the emitted electrons must be  $2.475 - 2.1 = 0.375 \text{ eV}$ . Or to put it another way, the voltage in the circuit is 0.375 volts. We therefore have  $1.3 \text{ W} = I(0.375)$ , or  $I = 3.467 \text{ amps} = (3.467)(6.24 \times 10^{18}) = 2.16 \times 10^{19} \text{ electrons/s}$ .

The cell's efficiency is  $2.16/2.46 = 87.8\%$